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REACTIONS INVOLVING A LIMITING REACTANTS

LIMITING REACTANT LIMITING REACTANTS In a given reaction there is not enough of one reagent to use up the other reagent completely. The reagent in short supply LIMITS the quantity of product. Reactants Products that can be formed.  $2 \text{ NO} + \text{O}_2 \rightarrow 2 \text{ NO}_2$  Limiting reactant Excess reactant LIMITING REACTANTS LIMITING REACTANTS Reaction to be Studied React solid Zn React solid Zn with 0.100 mol with 0.100 mol HCl a...

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2013 Fall CHEM101 Ch4 Review Worksheet Modified by Dr. Cheng-Yu Lai, 1) Global warming is thought to be caused by the increase of one particular gas. Name the gas. A) oxygen B) carbon monoxide C) carbon dioxide D) nitrogen E) helium Answer: C 2) How many grams of LiCl can be formed from 1.0 moles of Li? Assume an excess of Chlorine.  $2 \text{ Li(s)} + \text{Cl}_2(\text{g}) \rightarrow 2 \text{ LiCl(s)}$  A) 42.44 g LiCl B) 6.93 g LiCl C) 84.88 g LiCl D) ...

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Name: Chapter 20 and 21 Review Period: Chapter 20 and 21 Review Vocabulary - Know the following vocabulary and how to apply them. 1. Endothermic A. Mass is never created or 1. Closed System A. An experimental setup that does destroyed, only transformed; not capture the products. mass of reactants must equal the mass of the products. 2. Open System B. The reactant that is the first to be 2. Exotherm...

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Assessment Ch. 9 Practice Test Chapter: Stoichiometry In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. 1. A balanced chemical equation allows one to determine the a. mole ratio of any two substances in the reaction. b. energy released in the reaction. c. electron configuration of all elements in the reaction. d. mechanism...

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