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2013Fall CHEM101 Ch4 Review Worksheet Modified by Dr. Cheng-Yu Lai,1) Global warming is thought to be caused by the increase of one particular gas. Name the gas.A) oxygenB) carbon monoxideC) carbon dioxideD) nitrogenE) heliumAnswer: C2) How many grams of LiCl can be formed from 1.0 moles of Li? Assume an excess of Chlorine. $2 \text{ Li(s)} + \text{Cl}_2(\text{g}) \rightarrow 2 \text{ LiCl(s)}$ A) 42.44 g LiClB) 6.93 g LiClC) 84.88 g LiClD)...

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[REACTIONS INVOLVING A LIMITING REACTANTS](#)

LIMITING REACTANT LIMITING REACTANTSIn a given reaction there is not enough of one reagent to use up the other reagent completelyThe reagent in short supplyLIMITS the quantity of product Reactants Products that can be formed $2 \text{ NO g} + \text{O}_2 \text{ g} \rightarrow 2 \text{ NO}_2 \text{ g}$ Limiting reactant Excess reactant LIMITING REACTANTS LIMITING REACTANTS Reaction to be StudiedReact solid Zn React solid Zn with 0 100 mol with 0 100 mol HCl a...

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More Stoichiometry $2\text{AgNO}_3 + \text{Cu} \rightarrow \text{Cu}(\text{NO}_3)_2 + 2\text{Ag}$ 1. How many moles of solid copper would be needed to make 7.6 moles of solid silver? 2. Using the reaction in question #1: If you begin with 0.692 moles of silver nitrate, how many grams of silver can you expect to produce? 3. Using the reaction in question #1: What mass of silver nitrate would be needed in solution to produce a solution with 67.8 gra...

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